

VIBRATING BED PYROLYSIS SYSTEM
WITH A PRETREATMENT BED
TO REMOVE OXYGEN

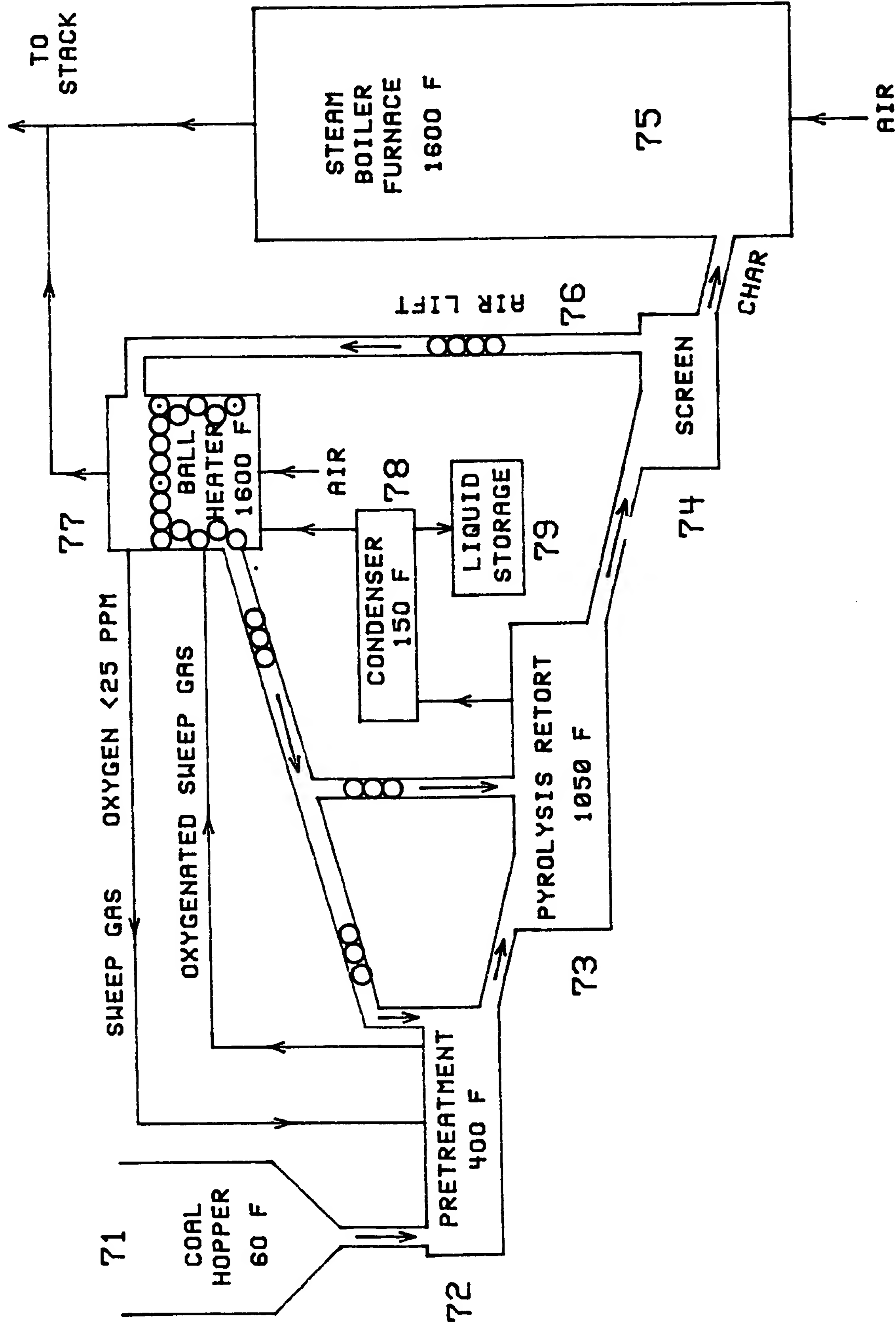
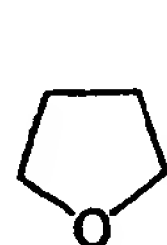
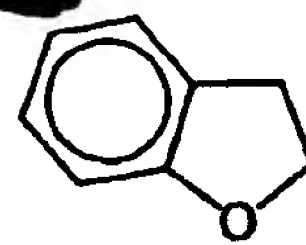


FIG. 2 COAL PRETREATMENT SYSTEM INCORPORATED IN A PYROLYSIS PROCESS UTILIZING CERAMIC BALLS FOR TRANSPORTING HEAT BETWEEN COMPONENTS

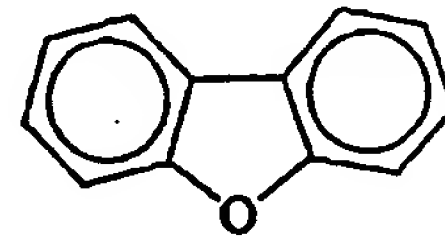
The most common for heterocyclic oxygen is in furan ring systems:



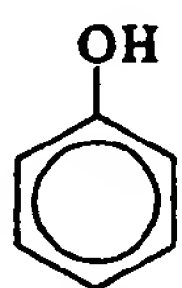
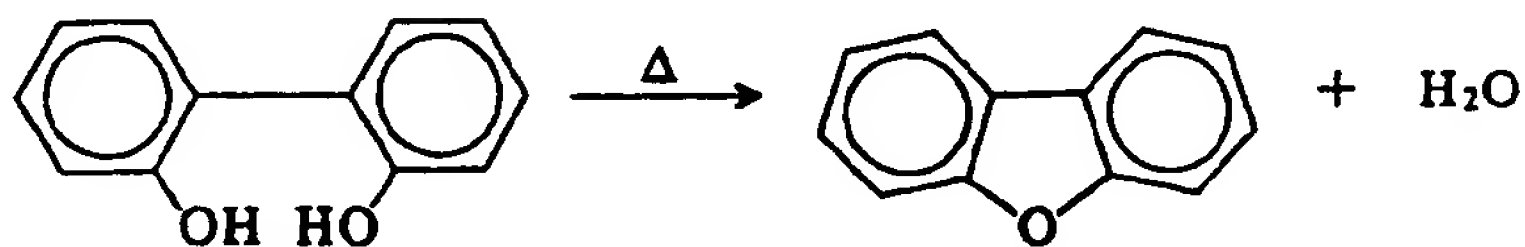
Furan



Benzofuran



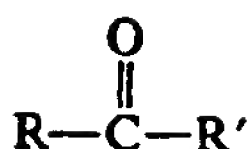
Dibenzofuran



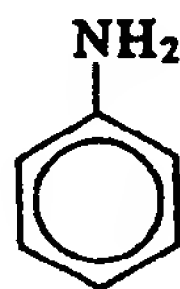
Phenol



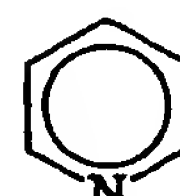
Carboxylic acid



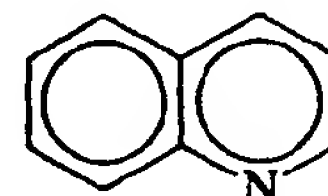
Carbonyl



Aniline



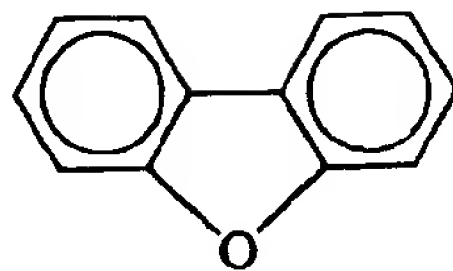
Pyridine



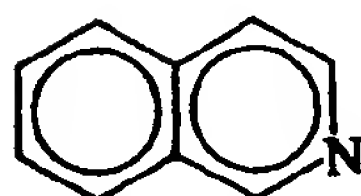
Quinoline



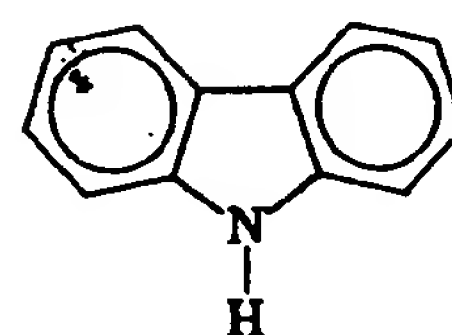
Ether



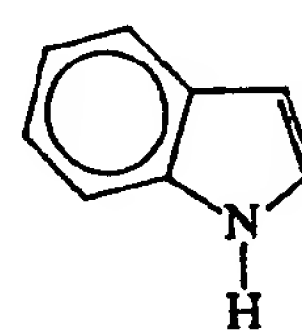
Heterocyclic oxygen



Isoquinoline



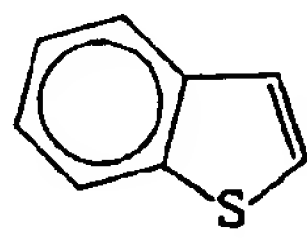
Carbazole



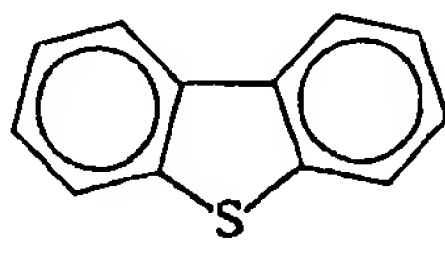
Indole



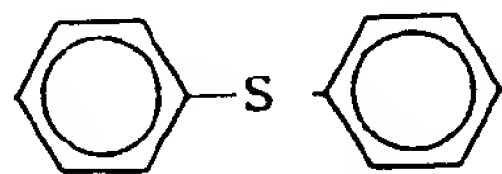
Thiophene



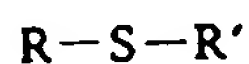
2,3-Benzothiophene



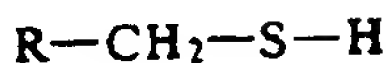
Dibenzothiophene



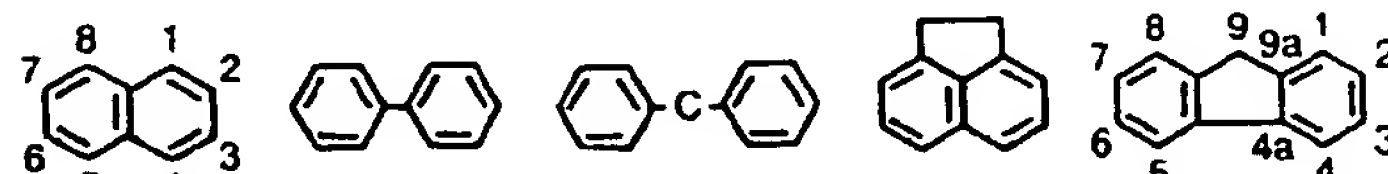
Diphenyl sulfide



Alkyl sulfide



Thiol



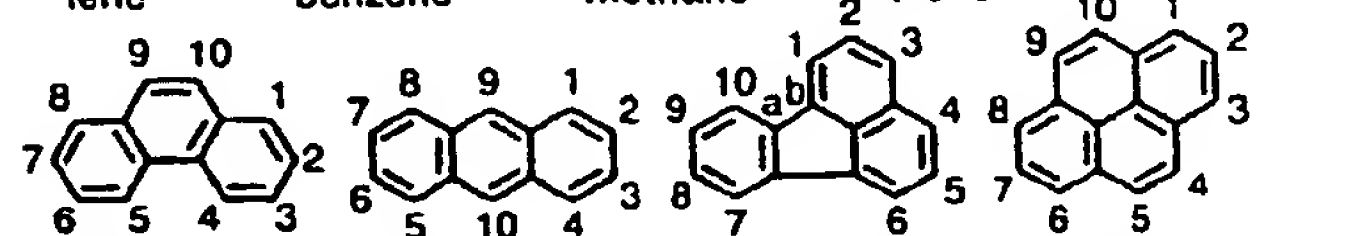
Naphthalene

Phenylbenzene

Diphenylmethane

Acenaphthene

Fluorene

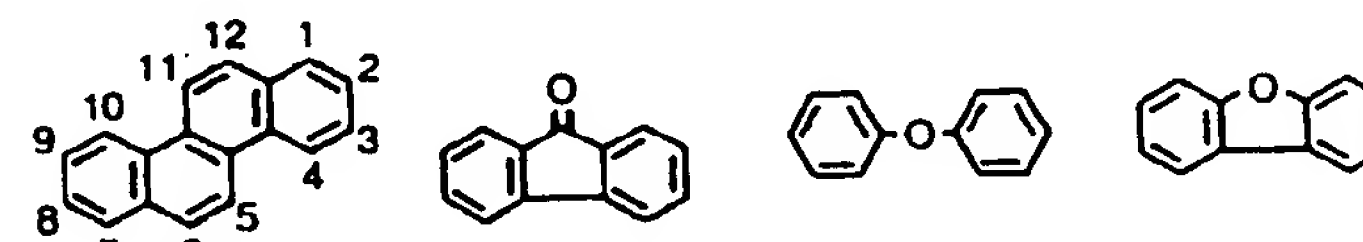


Phenanthrene

Anthracene

Fluoranthene

Pyrene

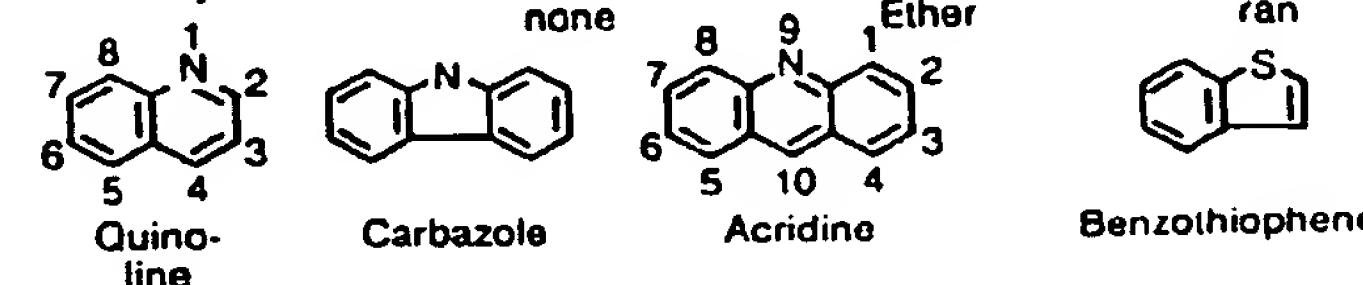


Chrysene

9-Fluorenone

Diphenyl ether

Dibenzofuran

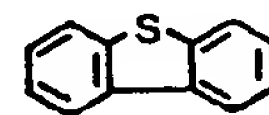


Quinoline

Carbazole

Acridine

Benzothiophene



Dibenzothiophene

FIG. 3A Structure of selected model compounds.

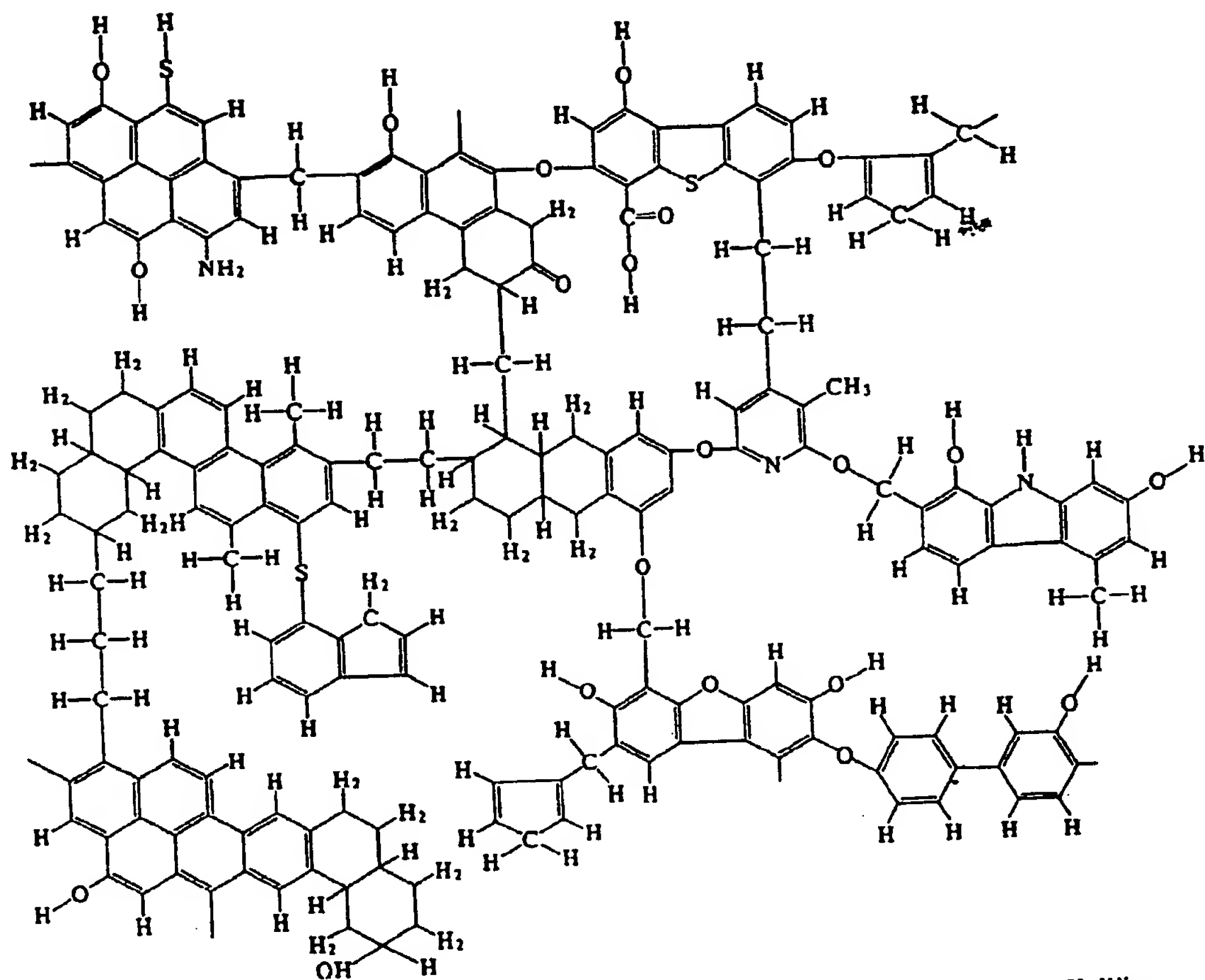


FIG. 3B Wisner model for bituminous coal. (Source: Reprinted with permission from W.H. Wisner, "Schematic Representation of Structural Groups and Connecting Bridges in Bituminous Coal," 1978.)

FIG. 4

GLASS SYSTEM PYROLYSIS TESTS

TEST NO. 10 - DEGASSING

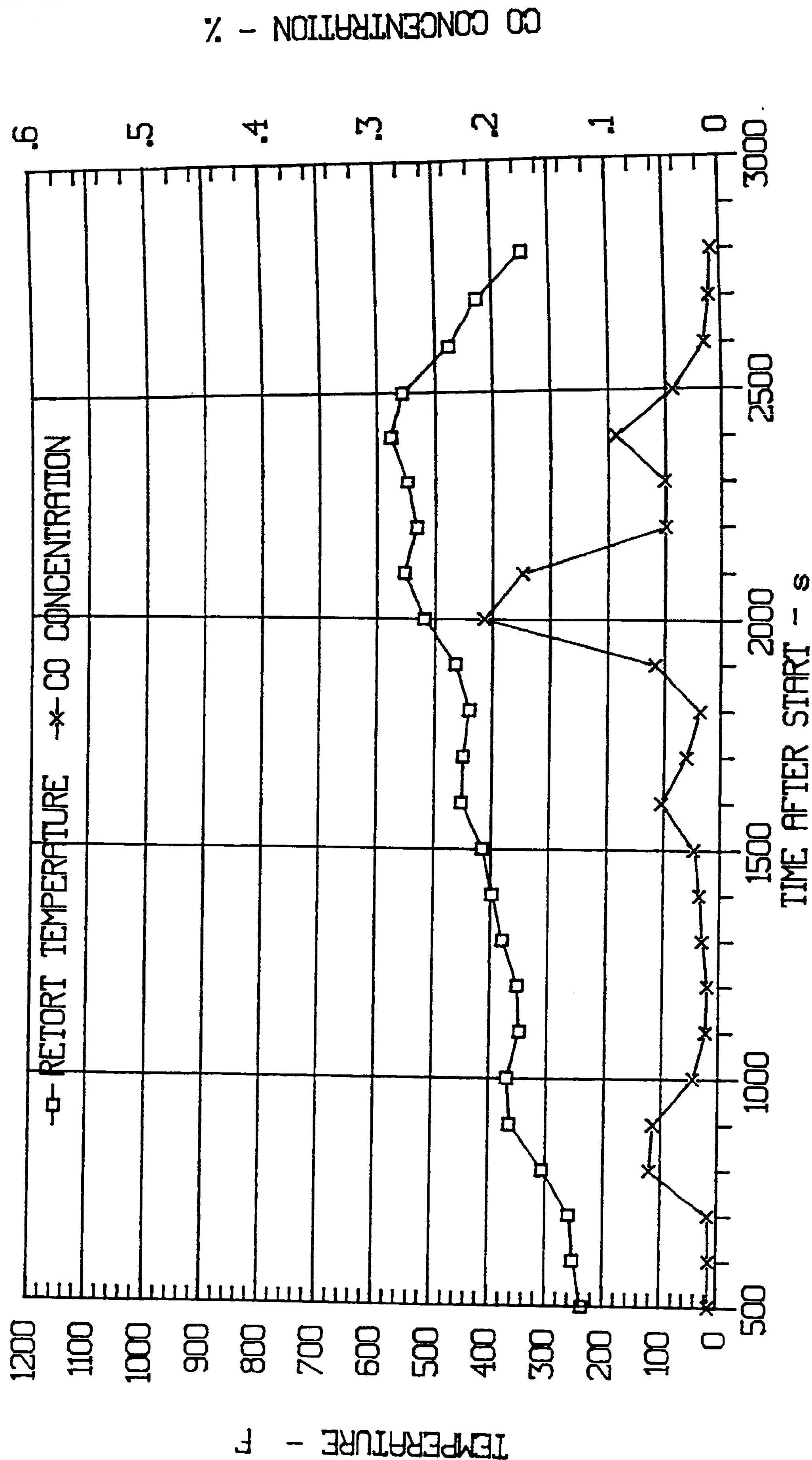
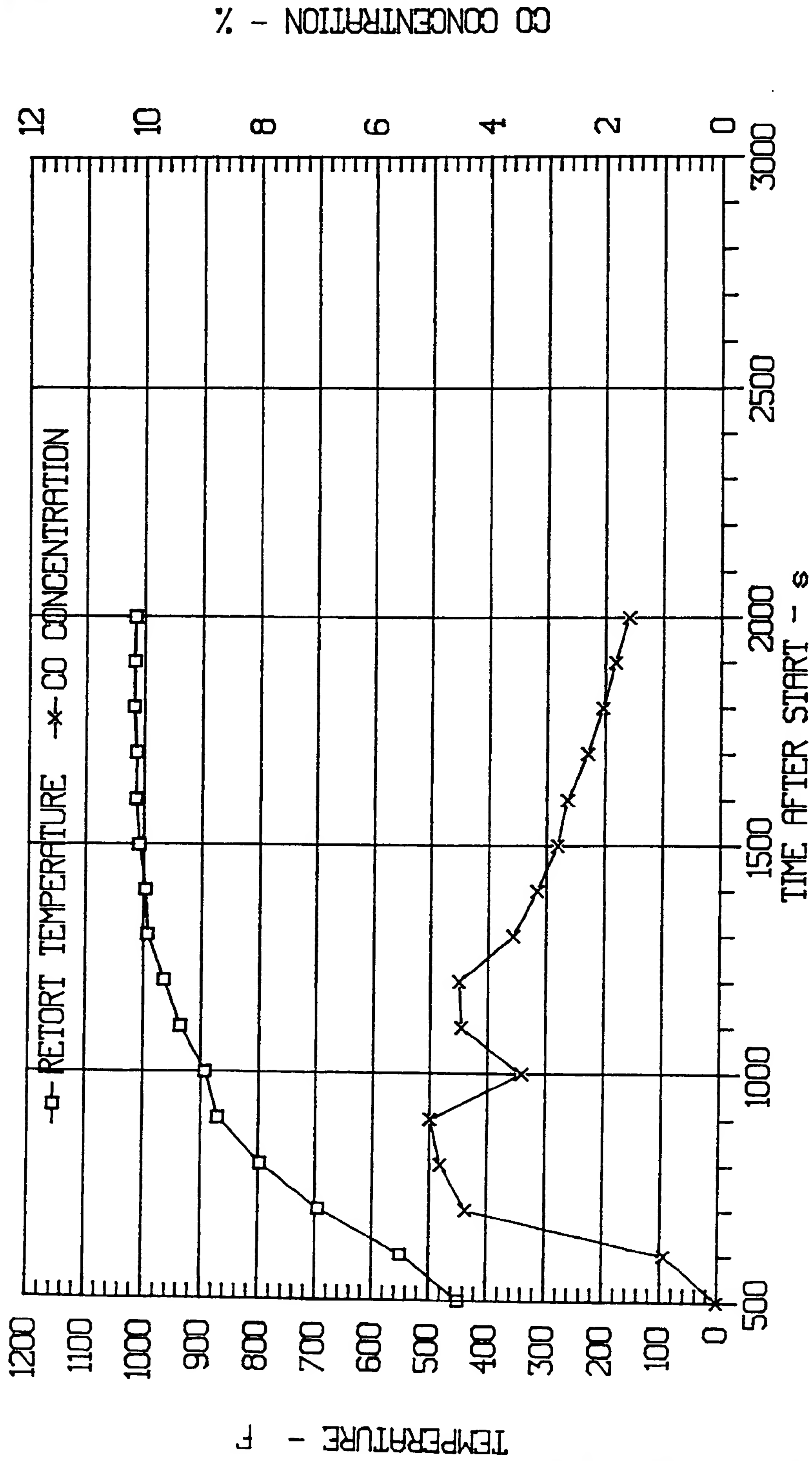


FIG. 5

GLASS SYSTEM PYROLYSIS TESTS

TEST NO. 10 - PYROLYSIS



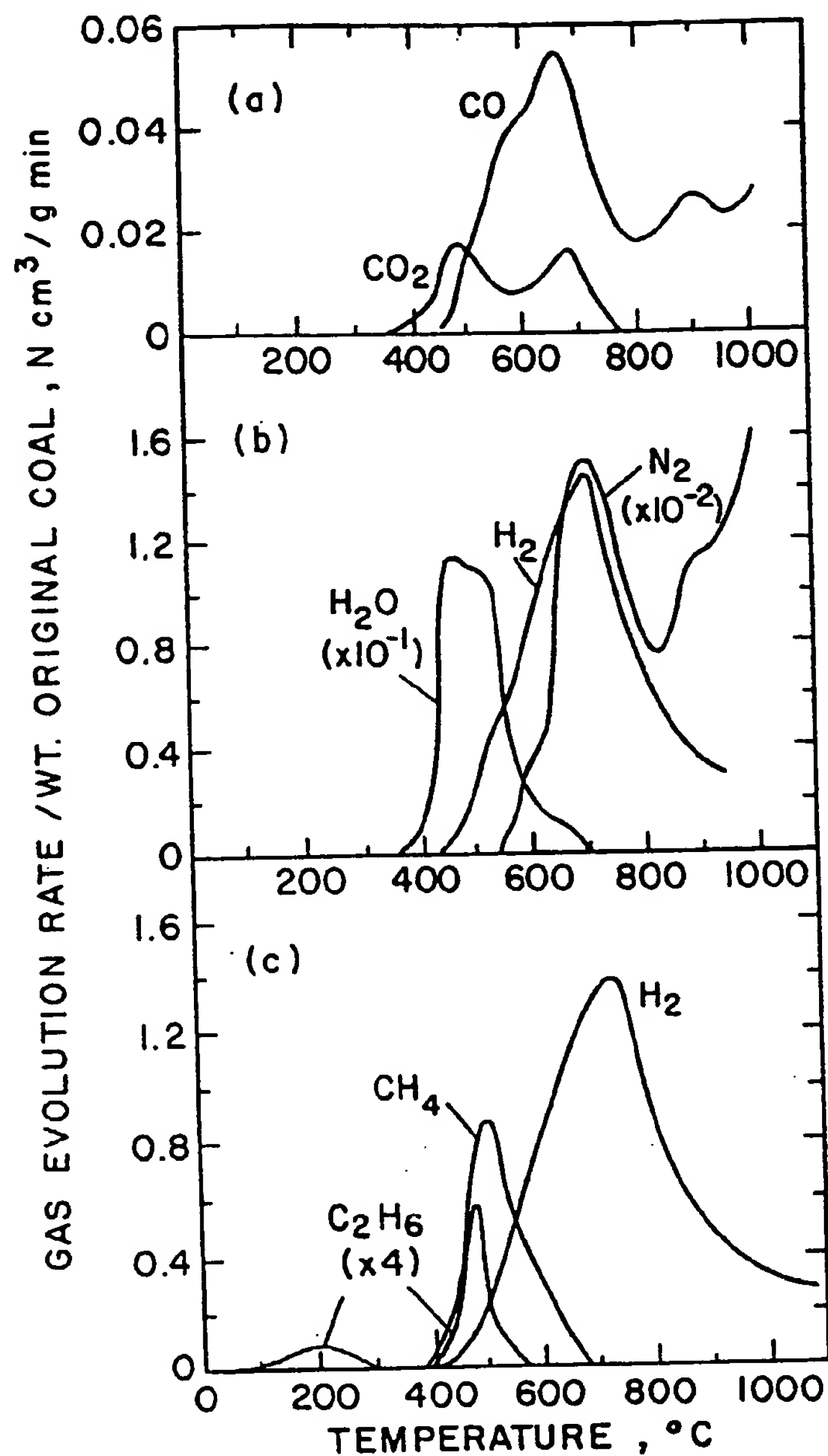


FIG. 6 Variation of Gaseous Species Evolution with Temperature during Coal Pyrolysis at Constant Heating Rate [(a) and (b), data of Klein (1971): Gustav coal, VM = 29 wt. % (MAF): heating rate = 1°C/min. (c), data of Jüntgen and Van Heek (1968): VM = 19.1 wt. %; heating rate = 2°C/min].

[illegible]

FIG. 7 Oxygen concentration in the off-gas and pretreatment bed temperature as functions of time from initiation of nitrogen purge.

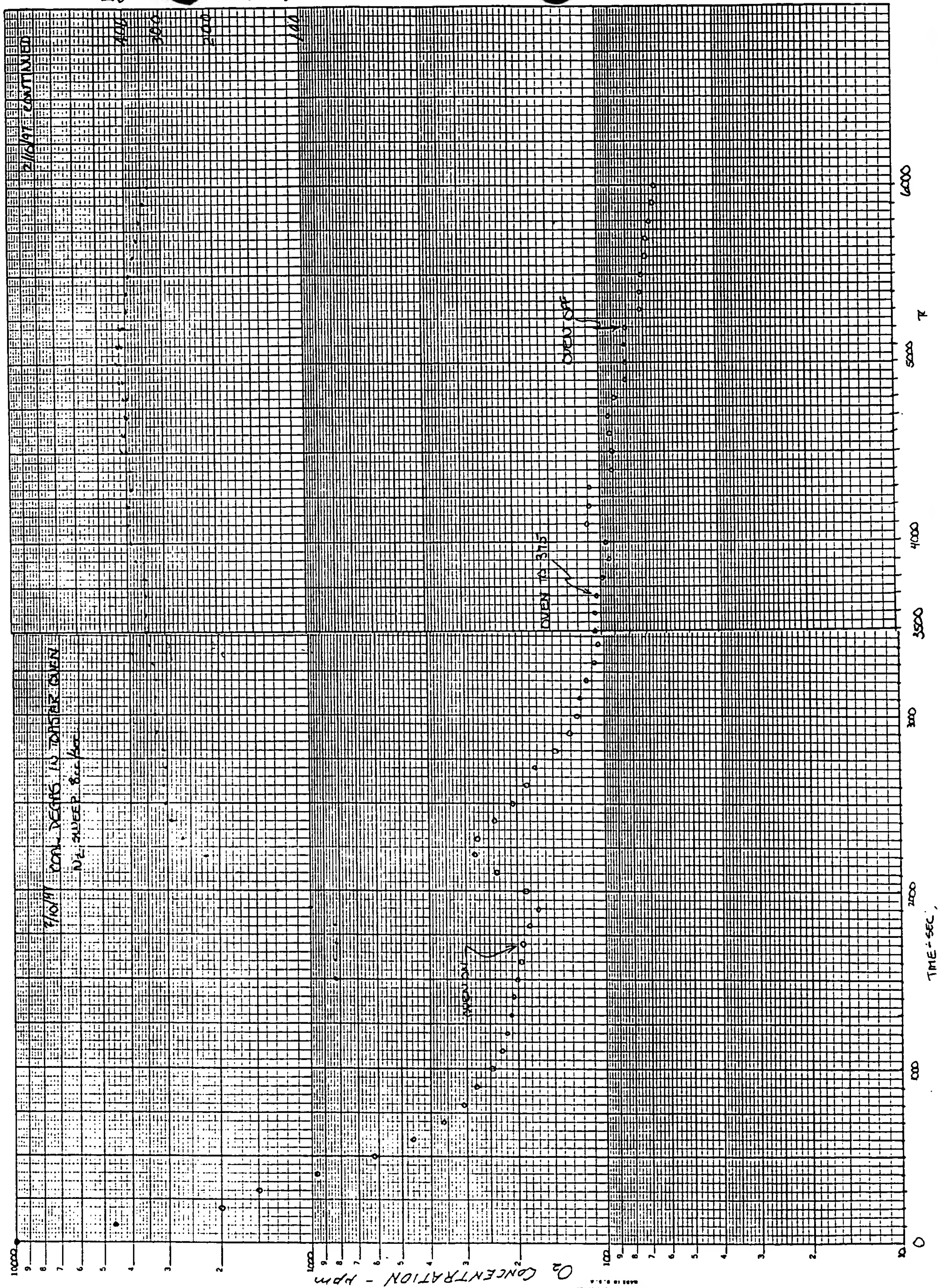


FIG. 8 Mass Spectrograph for pyrolysis product liquid after pretreatment at 450°F to remove oxygen, (Courtesy NREL)

01025637: Scan Avg 8-13 (0.99 - 1.62 min) - Back

